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### **Types Of Chemical Reactions Worksheet**

Types of Chemical Reaction Worksheet.  
Balance the reactions 1 to 6. and  
indicate which type of chemical reaction

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(synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented:  $2\text{NaBr} + \underline{1}\text{Ca(OH)}_2 \rightarrow \underline{1}\text{CaBr}_2 + \underline{2}\text{NaOH}$  Reaction Type : Double replacement.  $2\text{NH}_3 + \underline{1}\text{H}_2\text{SO}_4 \rightarrow \underline{1}(\text{NH}_4)_2\text{SO}_4$  Reaction Type : Synthesis.

## Types of Chemical Reaction

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Types of Chemical Reactions Answers  
Balance each of the following reactions  
and identify each type of reaction: 1.  $2 \text{NaBr} + \text{Ca(OH)}_2 \rightarrow \text{CaBr}_2 + 2 \text{NaOH}$   
double displacement 2.  $2 \text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4$  synthesis 3.  $4 \text{C}_5\text{H}_9\text{O} + 29 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$   
combustion 4.  $3 \text{Pb} + 2 \text{H}_3\text{PO}_4 \rightarrow 3 \text{H}_2 +$

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Pb 3 (PO 4) 2 single displacement 5. Li 3  
N + 3 ...

### **Types of Chemical Reaction Worksheet**

Six Types of Chemical Reaction  
Worksheet Answers Balance the  
following reactions and indicate which of  
the six types of chemical reaction are

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being represented: a)  $2 \text{NaBr} + 1 \text{Ca(OH)}_2 \rightarrow 2 \text{NaOH} + 1 \text{CaBr}_2$  Type of reaction:  
double displacement b)  $2 \text{NH}_3 + 1 \text{H}_2\text{SO}_4 \rightarrow 1 (\text{NH}_4)_2\text{SO}_4$  Type of reaction:  
synthesis c)  $4 \text{C}_5\text{H}_9\text{O} + 27 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$

**Balance the reactions a to e and  
indicate which types of ...**



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Types of Chemical Reaction Worksheet.  
Balance the reactions 1 to 6. and  
indicate which type of chemical reaction  
(synthesis, decomposition, single-  
displacement, double-displacement or  
combustion) is being represented:

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This 6-tab interactive flip book describes 6 main types of chemical reactions including synthesis, decomposition, single displacement, double displacement, combustion and acid-base reactions. The interactive tool provides descriptions, examples, analogies, pictures, and general reaction forms. This

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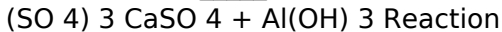
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Types of Reactions Worksheet Indicate the type of reaction taking place: 1)



Reaction type: \_\_\_\_\_ 2)  $\text{Ca(OH)}_2 + \text{Al}_2$



Reaction type: \_\_\_\_\_ 3)  $\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow \text{Fe} + \text{MgO}$

Reaction type: \_\_\_\_\_ 4)  $\text{C}_2\text{H}_4 + \text{O}_2$

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### **Alabama School of Fine Arts**

The chemical equation also does not give any information about the speed of the reaction. Some chemical equations and reactions have diverse affect.

Students likely find difficulty in balancing chemical equations worksheet. To help you resolve this issue, we have balancing equations worksheet with



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Answers. Balance the following reactions and indicate which of the six types of chemical reaction are being represented: 1)  $2. \text{NaBr} + \text{Ca(OH)}_2 \rightarrow \text{CaBr}_2 + 2. \text{NaOH}$ . Type of reaction: double replacement. 2)  $2. \text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4$ . Type of reaction: synthesis. 3) 4.  $\text{C}_5\text{H}_9\text{O} + 27.$

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### **Six Types of Chemical Reaction Worksheet**

Compilation of the 5 Types Chemical Reactions. Word equations included for all reactions. UPDATE: Synthesis Rxn-  
Word Equation: Iron(II) + Sulfur yields Iron(II) Sulfide

### **5 Types of Chemical Reactions Lab**

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Worksheets are Chemical reactions name, Types of reactions work, Balance the reactions a to e and indicate which types of, Balance the reactions 1 to 6, Six types of chemical reactions work, Types of chemical reactions, Work writing and balancing chemical reactions, Classifying chemical reactions

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Types of Chemical Reaction Worksheet  
Balance the reactions 1 to 6 and indicate  
which type of chemical reaction  
(synthesis, decomposition, single-  
displacement, double-displacement or



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combustion) is being represented: 1.

## **Types of Chemical Reaction Worksheet**

Six Types of Chemical Reaction  
Worksheet Answers Balance the  
following reactions and indicate which of  
the six types of chemical reaction are  
being represented: a)  $2 \text{NaBr} + 1 \text{Ca(OH)}$

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2 a)  $1 \text{ CaBr}_2 + 2 \text{ NaOH}$  Type of reaction:  
double displacement b)  $2 \text{ NH}_3 + 1 \text{ H}_2\text{SO}_4$   
 $\rightarrow 1 (\text{NH}_4)_2 \text{SO}_4$  Type of reaction:  
synthesis c)  $4 \text{ C}_5\text{H}_9\text{O} + 27 \text{ O}_2 \rightarrow 20$   
 $\text{CO}_2$  ...

## **Types of Chemical Reaction Worksheet answers - Six Types ...**

This worksheet covers six types of

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chemical reactions; synthesis, decomposition, single displacement, double displacement, combustion, and acid-base. The students are to identify each type of chemical equation by looking at the balanced equation. Then students are to justify their answer in the space

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