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4.2 For some hypothetical metal, the equilibrium number of vacancies at 900 C is $2.3 \times 10^{25} \text{ m}^{-3}$. If the density and atomic weight of this metal are 7.40 g/cm^3 and 85.5 g/mol , respectively, calculate the fraction of vacancies for this metal at 900 C.

Solution This problem is solved using two steps: (1) calculate the total number of lattice

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Fundamental Concepts
Electrons in Atoms. 2.1

Cite the difference
between atomic mass
and atomic weight.

Solution Atomic mass
is the mass of an
individual atom,
whereas atomic weight
is the average
(weighted) of the

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atomic masses of an atom's naturally occurring isotopes.

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Exercises with worked solutions Exercise E2.1

Material properties from experience. List the six main classes of engineering materials.

Use your own experience to rank them approximately:
(a) by stiffness (modulus, E). A sheet of a material that has a

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