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match Atomic radius  
Decrease Electron  
affinity  
electronegativity First  
ionization energy  
Increase Ionization  
energy Metals Noble  
gas configuration  
Noble gases Nonmetals  
Semimetal Shielding

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effect 1. Ionization energy is the energy required to remove an electron from an atom.  
2.

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Periodicity - page 6 G.  
Concept Mastery  
Questions 1. The  
shielding effect  
increases with  
increasing atomic

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number within a \_\_\_\_\_. a.  
period b. group c. both  
d. neither 2. In any \_\_\_\_\_,  
the number of  
electrons between the  
nucleus and the outer  
energy level is the  
same. a. period b.  
group c. both d. neither  
3.

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table 1. Which are  
metals? Circle your  
answers: C, Na, F, Cs,  
Ba, Ni Which metal in  
the list above has the  
most metallic  
character? Explain.  
Cesium - as the

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table 1. Which are  
metals? Circle your  
answers: C, Na, F, Cs,  
Ba, Ni Which metal in  
the list above has the  
most metallic  
character? Explain.

Cesium - as the largest  
atom, the lowest  
ionization energy and  
the most reactivity  
with nonmetals. This

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Chemistry: The

Periodic Table and

Periodicity Directions:

Answer each of the following questions.

You need not use complete sentences. 1.

Who first published the classification of the elements that is the basis of our periodic table today? DMITRI

MENDELEEV 2. By what

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property did

Periodicity  
Mendeleev arrange the

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metals that are more

reactive than the

transition elements

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- Ionization 3 4 Na-4.

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ANSWERS Directions:  
Answer the questions  
with the proper  
information using your  
notes, book, and the  
periodic table. 1.

Define a family.

GROUPS ON THE TABLE  
ARE SOMETIMES

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A. Periodic table 1.

Which are ...

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### **Answer Key**

During a chemical reaction, chemical bonds are 3. Use the bonding models to explain why. Scps Chemistry Worksheet Periodicity A Periodic Table 1 Which Periodic

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trends:

Electronegativity

answers. Name

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## Answers CHAPTER NOTES – CHAPTER 14 Chemical Periodicity

Goals : To gain an understanding of : 1 ...  
(1+) 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 0  
(none ... SCPS

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Periodicity 7. Give the names and chemical symbols for the elements that correspond to these atomic numbers: a. 10 b. 18 c. 36 d. 90 8. List, by number, both the ...

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adventure as well as  
experience

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**Table Topic 2**

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Periodicity - page 4 D.

Definitions - match

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Atomic radius

Decrease Electron

affinity

electronegativity First

ionization energy

Increase Ionization

energy Metals Noble

gas configuration

Noble gases Nonmetals

Semimetal Shielding

effect 1.

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answer key. Students fill in the missing information for the listed elements from the periodic table. ...

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a. C na f cs ba ni which metal in the list above has the most metallic character.

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**Answer Key**

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answer key or write the  
answer choices in a  
word bank for students  
to use. Its an easy 1  
word answer. Mass  
relationships in  
chemical reactions  
answers multiple  
choice questions. 6  
what specific group are  
the group 7 elements.

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