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### **Acid Neutralization Lab Report Answers**

Consider a 0.238 M aqueous solution of sodium hydroxide, NaOH. a) How many grams of NaOH are dissolved in 23.46 mL? b) How many individual hydroxide ions (OH<sup>-</sup>) are found in 23.46 mL? c) How many moles of sulfuric acid, H<sub>2</sub>SO<sub>4</sub>, are neutralized by 23.46 mL of 0.238 M NaOH(aq)? [Hint: begin by writing a balanced equation for this neutralization reaction.]

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Lab Manual Unit 28 Post-Lab Quiz Question 13 Part A What structure transports bile from the liver to the bile duct? ANSWER: Correct Lab Manual Unit 28 Post-Lab Quiz Question 12 Part A What cell produces bile? ANSWER: Correct Lab Manual Unit 28 Pre-Lab Quiz Question 6 Part A Which accessory organ produces secretions that moisten food? ANSWER: Correct Art-based Question Unit 28 Question 3 Part A ...

**10 of 35 1022017 751 PM Correct Lab Manual Unit 28 Post ...**

Question: Name Section Post-Lab Question Supplement Laboratory # 23 - Specific Heat Of Metals 1) Water Is Often Used As A Heat Storage Medium, Particularly In Solar Energy Systems. Does It Perform That Function Extremely Well For Some Reason, Or Is It Used Just Because It Is Easily Obtained? What Properties Of Water Make It Suitable, And What Single Property ...

**Solved: Name Section Post-Lab Question Supplement Laborato ...**

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Complete the conversion table at the bottom. We will pretend that each drop of solution is 1 mL. Using your average number of drops of KMnO4 as the number of mLs, use the data to calculate the molarity of the FeSO4. \_\_\_ Mn+7 + \_\_\_ Fe+2 → \_\_\_ Mn+2 \_\_\_ Fe+3 \_\_\_ mLs 20 mLs 0.01 M ?

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